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set (233) Water is an amazing liquid that makes. plant and animal life on Earth Possible. In addition to water forming the Earth's oceans water, it also forms the.

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molecule, H<sub>2</sub>O. • Each O-H bond is highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values) • bond angle of water = 105°.

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surfactant. aqueous solution. solvent.  
the inward force or pull that tends to  
minimize the surface ar.... any  
substance that interferes with hydrogen  
bonding between wa.... is water that  
contains dissolved substances.

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Surface tension. The inward force, or pull that tends to minimize the surface area of a liquid. Surfactant. Any substance that interferes with the hydrogen bonding between water molecules and thereby reduces surface tension. Aqueous solution.

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### **Aqueous Systems Flashcards ...**

Chapter 15 - Water and Aqueous Systems - 15.3 Heterogeneous Aqueous Systems - 15.3 Lesson Check - Page 507:  
21 Answer Colloids cannot be separated through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

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### **Chapter 15 - Water and Aqueous Systems - 15.3 Heterogeneous ...**

Chapter 15 "Water and Aqueous Systems" Pre-AP Chemistry Charles Page High School Stephen L. Cotton 2. Section 15.1 Water and its Properties □

OBJECTIVES: -Explain the high surface tension and low vapor pressure of water in terms of the structure of the water

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molecule and hydrogen bonding. 3.

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Title: Chapter 15 Review Water and Aqueous Systems 1 Chapter 15 Review Water and Aqueous Systems. Pre-AP Chemistry ; Charles Page High School ; Stephen L. Cotton; 2 Chapter 15 Review.



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Surface tension is the \_\_\_\_\_. How does the surface tension of water compare with the surface tensions of most other liquids? Which type of mixture(s) exhibit the ...

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Water And Aqueous Systems Chapter 15

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Chemistry . ... In a full glass of water, the water surface seems to bulge over the rim of the glass. Water forms nearly spherical drops at the end of an eyedropper. An insect called a water strider is able to "walk" on water. A. Those sentences describe how detergent molecules strengthen hydrogen bonding ...

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## **Water And Aqueous Systems Chapter 15 Chemistry - ProProfs Quiz**

Chapter 15 "Water and Aqueous Systems". use these activities to yourself study the vocabulary and major concepts presented in this chapter. heterogeneous mixture containing

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particles that are small enough to remain dispersed in the solvent and do not separate on standing.

## **Quia - Chapter 15 "Water and Aqueous Systems"**

Chapter 15 Water and Aqueous Systems159. SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449) This

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section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water. Water in the Liquid State (pages 445-447)

## **SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)**

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**Chapter 15 (&13.2) - Water & Aqueous Systems - KEIO ...**



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The high surface tension of water is due to the: a. small size of water molecules. b. low mass of water molecules. c. hydrogen bonding between water molecules. d. covalent bonds in water molecules. \_\_\_\_\_ 12. Salts and other compounds that remove moisture from air are said to be: a. efflorescent. c. colloidal. b. surfactant. d. hygroscopic.

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**05 CTR ch15 7/12/04 8:14 AM Page  
387 WATER AND AQUEOUS ...**

Chapter 15 Chapter 17 in your  
books“Water and Aqueous Systems” 2.  
Section 15.1 Water and its  
PropertiesOBJECTIVES:• Explain the high  
surface tension and low vapor pressure

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of water in terms of the structure of the water molecule and hydrogen bonding. •

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this, you should know how to avoid water wastage. However, there are many systems that do not require cleaning on installation. But, you must ensure that you clean the system on a regular basis.

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