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It is a nitrate with chemical formula KNO3. ... The solubility of Potassium Nitrate lab report. ... As per the given graph, the solubility of potassium nitrate at 60oC= 110 g and at 50oC = 91 g. Hence, by cooling a saturated solution of potassium nitrate containing 100 g of water from 60oC to 50oC, 19 g of the solute (110 g ...

Solubility Curve Of Potassium Nitrate Lab Answers

The solubility of Potassium Nitrate lab report. Topics: Solubility ... It is a nitrate with chemical formula KNO3. ... Molar solubility is defined as the amount (in moles) of solute per liter of saturated solution. The solubility of one compound ...

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Pre-Lab Questions 1. How do you determine whether a solution is saturated or not? 2. The equilibrium of a saturated LiCl aqueous solution is shown below: $\text{LiCl (s)} \rightleftharpoons \text{Li}^+ \text{(aq)} + \text{Cl}^- \text{(aq)}$ At 20°C, the solubility of LiCl in water is 550.0 g/L. (a) Calculate the molar concentration of LiCl. (b) Calculate the molar concentration of Li^+ and Cl^- .

THE THERMODYNAMICS OF POTASSIUM NITRATE DISSOLVING IN ...

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THE THERMODYNAMICS OF POTASSIUM NITRATE DISSOLVING IN WATER1 OBJECTIVE In this experiment, the changes in free energy (ΔG), enthalpy (ΔH), and entropy (ΔS) of the potassium nitrate (KNO3) dissolving reaction will be determined by measuring the equilibrium constant (K sp) at different temperatures.

THE THERMODYNAMICS OF POTASSIUM NITRATE DISSOLVING IN WATER1

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